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 $3Fe + conc^n 8HNO_3 \longrightarrow Fe_3O_4 + 4H_2O + 8NO_2$ 22.(b) 23.(a) ZnO is amphoteric oxide but other are basic 24.(c) $2NaOH + Zn \rightarrow Na_2ZnO_2 + H_2$ 25.(b) Na is highly electropositive metal 26.(c) $\Delta n = 1 - 4 = -3$ $K_p = K_c (RT)^{\Delta n}$ $K_p = K_c (RT)^{-3}$ $K_p = \frac{K_c}{(RT)^3}$ $\therefore K_c > K_p$ 27.(b) 28.(c) 29.(a) Total no. of elements = 5. No. of subsets having not more than 3 elements = c(5, 3) + c(5, 2) + c(5, 1) + c(5, 0)= 10 + 10 + 5 + 1 = 16 $2\tan^2 x = \sec^2 x$ 30.(d) i.e. $2\tan^2 x = 1 + \tan^2 x$ i.e. $\tan^2 x = 1$ i.e. $\tan^2 x = \tan^2 \frac{\pi}{4}$ $\therefore x\pi \pm \frac{\pi}{4}$ 31.(d) We have $\frac{a}{\sin A} = \frac{b}{\sin B}$ i.e. $\frac{3}{3} = \frac{4}{\sin B}$ i.e. sinB = 1 $\therefore B = 90^{\circ}$ 32.(c) $\vec{a} = \lambda \hat{a}$ i.e. $|\vec{a}| = \lambda |\hat{a}|$ i.e. $\lambda = |\vec{a}|$ 33.(b) $1 - \log_e 2 + \frac{(\log_e 2)^2}{2} - \frac{(\log_e 2)^3}{3!} + \frac{(\log_e 2)^3}{3!}$ $=e^{\log_{e}\frac{1}{2}}=\frac{1}{2}$ 34.(d) Sum of roots = 0i.e. $-\frac{\lambda-5}{\lambda-2}=0$ i.e. $\lambda = 5$ 35.(b) We have |A. Adj A| = |A| |I| = 8.1 = 836.(b) Here D = $\begin{vmatrix} 1 & 3 \\ 4 & -1 \end{vmatrix} = -1 - 12 = -13 \neq 0$ So it has unique solution. 37.(c) The required line is 5(x-2) - 3(y-3) = 0i.e. 5x - 3y - 19 = 038.(d) y-coordinates of centroid = 0 i.e. $\frac{a+b-3}{3} = 0$ i.e. a + b = 3

39.(b) By definition 'b'
40.(a) Here a² = 25, b² = 9, e =
$$\sqrt{1 - \frac{b^2}{a^2}} = \sqrt{1 - \frac{9}{25}}$$

 $= \sqrt{\frac{16}{25}} = \frac{4}{5}$
41.(d) Here $16x^2 - 9y^2 = 144$
i.e. $\frac{x^2}{9} - \frac{y^2}{16} = 1$
So $|PS_1 - PS_2| = 2.3 = 6$
42.(d) By definition, option 'd' is correct.
43.(b) $x \to 0$ $\frac{\sin x - \sin bx}{x} = \lim_{x \to 0} \left[\frac{\sin ax}{ax} \cdot a - \frac{\sin bx}{bx} \cdot b \right]$
 $= 1.a - 1.b = a - b$
44.(d) $\lim_{x \to 0} f(x) = f(0)$
i.e. $\cos 0 + 1 = K$ i.e. $K = 2$
45.(a) $e^{3\log_2 x} = x^3$ and its derivative is $3x^2$
46.(c) $f'(x) = 2x - 4$
So, $f'(x) > 0$
i.e. $2x - 4 > 0$
i.e. $2x - 4 > 0$
i.e. $x > 2$
47.(b) Put 1 + e^x = y
i.e. $dy = e^x dx$
Then $\int \frac{e^x dx}{1 + e^x} = \int \frac{dy}{dx} = \ln y + c = \ln(1 + e^x) + c$
48.(b) Req. area $= \int_{-1}^{2} dx = \int_{-1}^{2} x^3 dx = \left[\frac{x^4}{4}\right]_{-1}^{2} = \frac{16}{4} - \frac{1}{4}$
 $= \frac{15}{4}$ sq. units
49. (a) 50. (a) 51. (d) 52. (c) 53. (b) 54. (b)
55. (c) 56. (a) 57. (a) 58. (b) 59. (b) 60. (b)
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61.(b) $a_t = 2m/s^2$
 $a_c = \frac{v^2}{r} = \frac{30^2}{500} = \frac{9}{5} = 1.8 m/s^2$
 $\therefore a = \sqrt{a_c^2 + a_t^2} = \sqrt{1.8^2 + 2^2}$
62.(c) $2t_s = t_r$
or, $2\sqrt{\frac{21}{gsin\theta}} = \sqrt{\frac{21}{gsin\theta - \mu gcos\theta}}$
or, $4\sin\theta - 4\mu\cos\theta = \sin\theta$
or, $4in\theta = \frac{1}{\sin\theta - \mu\cos\theta}$
or, $4in\theta = 4\mu\cos\theta$

or,
$$200 = (l^2 \times 2) \times 1$$

or, $l = 10$ cm

PEA Association Pvt. Ltd. Thapathali, Kathmandu, Tel: 4245730, 4257187 2078-05-26 Hints & Solution Gain in time in day = $\frac{1}{2} \propto 4\theta \times 1$ day 71.(b) $I = I' + 2\sqrt{I'I'}\cos\phi + I'$ 64.(c) if path difference is $\boldsymbol{\lambda}$ then phase difference is or, $\alpha = \frac{8.6 \times 2}{10 \times 86400} = 2 \times 10^{-5} / ^{\circ} C$ 2π so I = 4I' \Rightarrow I' = $\frac{1}{4}$ 65.(c) $Q = \frac{kAd\theta}{2l} \times t_1 = \frac{k2Ad\theta}{l} \times t_2$ If path difference is $\frac{\lambda}{4}$ then phase difference (ϕ ') or, $\frac{t_1}{2} = 2t_2$ or, $t_2 = \frac{12}{4} = 35$ $=\frac{2\pi}{\lambda}\frac{\lambda}{4}=\frac{\pi}{2}$ 66.(b) $\rho_m = \frac{4V\rho_H + V\rho_0}{4V + V} = \frac{4\rho_H + \rho_0}{5}$ $I_R = I' + 2\sqrt{I'I'} \cos\phi' + I'$ $= 2I' = 2 \times \frac{I}{4} = \frac{I}{2}$ $\therefore \quad \frac{v_{m}}{v_{H}} = \sqrt{\frac{\rho_{H}}{\rho_{m}}} = \sqrt{\frac{\rho_{H}}{4\rho_{H}}} = \frac{1}{2}$ 72.(a) :. $v_m = \frac{1224}{2} = 612 \text{ m/s}$ 67.(a) $V = \frac{Q}{4\pi\varepsilon_0 R} - \frac{Q}{4\pi\varepsilon_0 3R} = \frac{Q}{4\pi\varepsilon_0 R} \left(1 - \frac{1}{3}\right)$ $\therefore \frac{Q}{4\pi\epsilon_0 R} = \frac{3}{2} V$ Again, $E = \frac{Q}{4\pi\epsilon_0(3R)^2} = \frac{Q}{4\pi\epsilon_0 R} \times \frac{1}{9R}$ tanc = $=\frac{3V}{2} \times \frac{1}{9R}$ or, $R = h \frac{\text{sinc}}{\cos c}$ $= h \frac{\text{sinc}}{\sqrt{1 - \sin^2 c}}$ $= h \frac{\text{sinc}}{\sqrt{1 - \sin^2 c}}$ 68.(a) Resistance in GHF is open so $R_{eq} = \frac{4 \times 2}{4 + 2} + 2 = \frac{4}{3} + 2 = \frac{10}{3} \Omega$ $=h \times \frac{1}{\mu} \frac{1}{\sqrt{1-\frac{1}{\mu^2}}} = \frac{h}{\sqrt{\mu^2-1}}$ $I = \frac{E}{R_{eq}} = \frac{10}{10} = 3A$ $=\frac{4}{\sqrt{\left(\frac{5}{\sqrt{3}}\right)^2-1}}=4\times\frac{3}{4}$ Here current 1A flows along BGD & 2A along AD so voltage across capacitor = Pd across GP $= V_{GD} + V_{DF} = 1 \times 2 + 3 \times 2 = 8V$ 69.(c) $B = \frac{\mu I}{2R} = \frac{\mu_0 f}{2R}$ 1st case 73.(c) or, $6.28 = \frac{4\pi \times 10^{-7} \times 2 \times 10 \times 6.25 \times 10^{12}}{2R}$ $\frac{hc}{\lambda} = \phi + 3ev_0 \dots (i)$ or, R = $\frac{4\pi \times 10^{-7} \times 2 \times 10^{-6} \times 6.25 \times 10^{12}}{2 \times 6.28}$ 2nd case $\frac{hc}{2\lambda} = \phi + ev_0 \dots (ii)$ 70.(b) $\cos\phi = \frac{R}{Z} = \frac{R}{\sqrt{R^2 + (X_L - X_C)^2}}$ From (i) & (ii) $\frac{1}{2}(\phi + 3\mathrm{ev}_0) = \phi + \mathrm{ev}_0$ $=\frac{10}{\sqrt{10^{2}+\left(\omega L-\frac{1}{\omega c}\right)^{2}}}=\frac{10}{10\sqrt{2}}=\frac{1}{\sqrt{2}}$ or, $\phi + 3ev_0 = 2\phi - 2ev_0$ or, $\phi = ev_0 \dots (iii)$ From equation (i) $= \tan 45^{\circ}$ $\frac{hc}{\lambda} = \phi + 3\phi$ $\phi = 45^\circ = \frac{\pi}{4}$

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	or, $4\phi = \frac{hc}{\lambda}$
	or, $4\frac{hc}{\lambda_0} = \frac{hc}{\lambda}$
	or, $\lambda = 4\lambda$
74.(d)	$\underline{N}_{A} = \underline{1}$
/ 1.(u)	100 /
	or, $\frac{N_0 e^{-10\lambda t}}{N_0 e^{-\lambda t}} = \frac{1}{e}$
	or, $\frac{1}{e^{9\lambda t}} = \frac{1}{e}$
	or, $\left(\frac{1}{e}\right)^{9\lambda t} = \frac{1}{e}$
	or, $9\lambda t = 1$
	or, $t = \frac{1}{9\lambda}$
75.(c)	$K_2Cr_2O_7 + 2KOH \longrightarrow 2K_2CrO_4 + H_2O$
	red orange lemon yellow
76.(d)	$Cl_2 + H_2O \longrightarrow HCl + [O]$
/ \	Colour vegetable + $[O] \rightarrow$ Colourless vegetable
77.(c)	KOU (ala) Pr. NaNHa/NHa
	$I \xrightarrow{\text{KOH (alc.)}} \text{CH}_3\text{CH} = \text{CH}_2 \xrightarrow{\text{Br}_2} \text{CH}_3 - \text{CH} - \text{CH}_2\text{Br} \xrightarrow{\text{NaNH}_2/\text{NH}_3} \text{CH}_3\text{C} = \text{CH}_2\text{Propyne}$
78.(b)	$\frac{\text{Wt. of metal}}{\text{EW of metal}} = \frac{\text{Wt. of oxygen}}{\text{EW of oxygen}}$
	$\frac{2}{x} = \frac{0.8}{8}, x = 20$
	valency of metal = 2
	At. wt. = EW × valency = $20 \times 2 = 40$
79.(d)	At. no. = $19 = 19e \& 19P$
	No. of $n = 39 - 19 = 20n$
80.(a)	1L of $\frac{N}{2}$ HCl contains 18.25g of HCl
	Remaining HCl = 18.25 – 16.425 = 1.825
	$\frac{W}{E} = \frac{N \times V_m}{1000}$
	E 1000
	$\frac{1.825}{36.5} = \frac{N \times 500}{1000}, N = 0.1N \text{ HCl}$
	$pH = -\log[H_3O^+]$
	$= -\log[0.1]$
	= 1
81.(a)	$CaC_2O_4 \Longrightarrow Ca^{++} + C_2O_4^{}$
	S S S
82.(a)	$f(x) = \frac{\cos^2 x + \sin^4 x}{\sin^2 x + \cos^4 x} = \frac{\cos^2 x + \sin^2 x (1 - \cos^2 x)}{\sin^2 x + \cos^2 x (1 - \sin^2 x)}$
	$=\frac{\cos^{2}x + \sin^{2}x - \sin^{2}x \cos^{2}x}{\sin^{2}x + \cos^{2}x - \sin^{2}x \cos^{2}x}$
	$=\frac{1-\sin^2 x \cos^2 x}{1-\sin^2 x \cos x}=1$
	$=\frac{1}{1-\sin^2 x \cos x}=1$

83.(c) Here $\alpha + \beta = \frac{5}{6}, \alpha\beta = \frac{1}{6}$ So $\tan^{-1}\alpha + \tan^{-1}\beta = \tan^{-1}\frac{\alpha+\beta}{1-\alpha\beta} = \tan^{-1}\frac{\frac{5}{6}}{1-\frac{5}{7}}$ $= \tan^{-1} 1 = \frac{\pi}{4}$ 84.(a) Projection of \vec{b} on $\vec{a} = \frac{\vec{a} \cdot \vec{b}}{|\vec{a}|} = \frac{2.5 + 1.-3 + 2.1}{\sqrt{2^2 + 1^2 + 2^2}}$ $= \frac{10 - 3 + 2}{\sqrt{9}} = \frac{9}{3} = 3$ 85.(a) No. of arrangements of CALCUTTA = $\frac{8!}{2! \; 2! \; 2!}$ = 5040 No. of arrangements of AMERICA = $\frac{7!}{2!}$ = 2520 So ratio of arrangements = 5040 : 2520 = 2:1We have, $(1 + x)^n = c_0 + c_1 x + c_2 x^2 + c_3 x^3 + ... + c_n x^n$ 86.(c) We have, $(1 + x) = c_0 + c_1 x + c_2 x + c_3 x + \dots + c_n x$ Putting x = 2, we get $(1 + 2)^n = c_0 + 2.c_1 + 2^2.c_2$ $+ 2^3.c_3 + \dots + 2^n c_n$ i.e. $c_0 + 2c_1 + 2^2.c_2 + 2^3.c_3 + \dots + 2^n.c_n = 3^n$ $\frac{1}{3.7} + \frac{1}{7.11} + \frac{1}{11.15} + \dots$ 87.(d) $=\frac{1}{4}\left[\left(\frac{1}{3}-\frac{1}{7}\right)+\left(\frac{1}{7}-\frac{1}{11}\right)+\left(\frac{1}{11}-\frac{1}{15}\right)+\ldots\right]$ $=\frac{1}{4}\cdot\frac{1}{3}=\frac{1}{12}$ 88.(b) Given $\left(\frac{1+i}{1-i}\right)^m = 1$ i.e. $\left(\frac{1+i}{1-i} \times \frac{1+i}{1+i}\right)^m = 1$ i.e. $\left(\frac{1+2i-1}{1+1}\right)^m = 1$ i.e. $i^m = 1 \implies m = 4$ 89.(b) The condition for one slope equals n times the other is $4\mathbf{n}\mathbf{h}^2 = \mathbf{a}\mathbf{b}(1+\mathbf{n})^2$ Put x = 5, $4.5 \cdot h^2 = ab \cdot (1 + 5)^2$ i.e. $20h^2 = 36 ab$ i.e. $5h^2 = 9ab$ 90.(a) Here vertex = (a, 0)i.e. h = a, K = 0Focus = $(a', 0) (b + \alpha, K) = (a', 0)$ i.e. $\alpha = a' - a$ So the equation of parabola is $(y-K)^2 = 4\alpha(x-h)$ i.e. $y^2 = 4(a' - a)(x - a)$

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